

# LAB ACTIVITY ROCK CORRELATION ANSWERS Workbook

## Lab Activity Rock Correlation Answers: An In-Depth Guide

**Lab activity rock correlation answers** are essential for students and geologists aiming to understand the relationships between different rock units across various geographic locations. Rock correlation involves comparing rock layers (strata) to determine if they are the same age, origin, or depositional environment. This process is fundamental in geology for reconstructing Earth's history, locating mineral deposits, and understanding geological processes. This article provides a comprehensive overview of how to approach rock correlation activities, interpret data, and find accurate answers during lab exercises.

## Understanding the Basics of Rock Correlation

### What is Rock Correlation?

Rock correlation is the method of matching rock layers from different locations based on their characteristics. It helps geologists identify continuous strata and understand the geological history of an area. Correlation can be based on various features including:

- Fossil content
- Mineral composition
- Color and texture
- Stratigraphic position
- Structural features
- Radiometric dating results

### Importance of Accurate Correlation

Correctly correlating rock layers allows geologists to:

- Reconstruct past environments
- Determine the relative ages of rocks
- Identify regional geological events
- Locate potential resource deposits

- Understand tectonic movements

## Common Methods Used in Rock Correlation

### Lithostratigraphic Correlation

This method involves comparing physical characteristics of rocks, such as mineral content, color, grain size, and layering. It is especially useful in sedimentary rocks.

### Biostratigraphic Correlation

Using fossils within the rocks, biostratigraphy compares fossil assemblages to establish age and correlation. Index fossils?species that existed during a short time frame?are particularly valuable.

### Chronostratigraphic Correlation

Based on radiometric dating, this approach assigns absolute ages to rock units, enabling precise correlation between distant locations.

### Magnetostratigraphic Correlation

Utilizes the Earth's magnetic field record preserved in rocks. Magnetic polarity reversals can serve as markers across different regions.

## Step-by-Step Approach to Lab Activity Rock Correlation

### 1. Examine and Document Rock Samples

Start with detailed observations of each rock sample or layer:

- Note physical features: color, texture, grain size
- Identify any fossils or mineral inclusions
- Record stratigraphic position within the local sequence

### 2. Collect and Analyze Fossil Data

If fossils are present, identify species and compare with known index fossils. This helps determine relative ages and regional correlations.

### 3. Compare Mineral Composition and Features

Use microscopy or chemical tests to analyze mineral content. Similar mineralogy indicates potential correlation.

#### **4. Use Radiometric Dating (If Available)**

Obtain absolute ages to anchor the relative timeline. Cross-reference these dates with fossil data and lithological features.

#### **5. Assess Structural and Stratigraphic Relationships**

Identify structural features such as faults, folds, or unconformities that may complicate correlation. Recognize lateral changes in rock units.

#### **6. Correlate Using Multiple Data Sets**

Combine lithostratigraphic, biostratigraphic, and chronostratigraphic data to make the most accurate correlations. Cross-verification increases confidence in your conclusions.

#### **7. Document and Justify Your Correlation**

Prepare a detailed report explaining the reasoning behind your correlations, supported by data and observations.

### **Common Challenges and How to Overcome Them**

#### **Difficulty in Fossil Identification**

Fossil preservation varies, making identification challenging. Use reference charts and compare with known fossil records. When fossils are absent, rely on lithology and mineralogy.

#### **Complex Tectonic Deformations**

Structural disturbances like folding or faulting can displace strata. Recognize these features and interpret their effects on stratigraphic continuity.

#### **Unconformities and Gaps**

Unconformities indicate erosion or non-deposition periods. Be cautious when correlating layers separated by unconformities; look for continuous markers such as fossils or radiometric dates elsewhere.

## Limited Data

Sometimes, samples lack fossils or radiometric data. In such cases, focus on lithological similarities and regional geological context.

## Sample Rock Correlation Answers and How to Interpret Them

### Example 1: Matching Sedimentary Layers

Suppose you observe two sedimentary layers from different sites. Both contain sandstone with similar grain size, color, and fossil content of a specific index fossil. Radiometric dates are also close. Based on this data, you conclude these layers are correlatable, representing the same depositional period.

### Example 2: Discrepant Data Sets

In another case, fossils differ significantly between two layers, but mineral composition and stratigraphic position suggest they are contiguous. This may indicate an unconformity or reworking of fossils. You might then decide to use radiometric dating or structural features to clarify correlation.

## Tips for Success in Rock Correlation Lab Activities

- Always record detailed observations with sketches and notes.
- Use multiple lines of evidence to support your correlation conclusions.
- Stay aware of geological processes that may complicate stratigraphic relationships.
- Consult reference materials and fossil identification guides.
- Practice interpreting different types of data and integrating them effectively.

## Conclusion

Understanding and mastering **lab activity rock correlation answers** requires a combination of careful observation, analytical skills, and knowledge of geological principles. Whether you are working through sedimentary, metamorphic, or igneous rocks, the key lies in integrating lithological, fossil, and chronological data to make accurate correlations. As you develop your skills, you'll become increasingly adept at reconstructing Earth's geological history, a fundamental goal in geology. Remember, accuracy in correlation not only advances your academic understanding but also contributes significantly to geological research and resource exploration.

Lab Activity Rock Correlation Answers: An In-Depth Review

Understanding rock correlation is fundamental in geology and earth sciences, especially when analyzing stratigraphic sequences, reconstructing geological histories, or conducting resource exploration. The lab activity focusing on rock correlation answers offers students and professionals an essential opportunity to develop skills in identifying, matching, and interpreting rock units across different outcrops or boreholes. This comprehensive review delves into the core concepts, methodologies, common challenges, and best practices associated with lab activities centered on rock correlation.

## Introduction to Rock Correlation

Rock correlation is the process of identifying and matching similar stratigraphic units across different geographic locations. It enables geologists to piece together the spatial and temporal relationships of rock layers, which is crucial for constructing geological maps, understanding depositional environments, and exploring natural resources.

Key Objectives of Rock Correlation:

- Establish continuity of stratigraphic units
- Recognize lateral variations and discontinuities
- Infer depositional environments and geological history
- Support resource exploration and environmental assessments

## Foundational Concepts in Rock Correlation

### Stratigraphy and Stratigraphic Units

- Definition: The study of layered rocks (strata) and their chronological relationships.
- Types of units: Formations, members, beds, and groups.
- Significance: Units are used as fundamental units for correlation.

### Principles of Stratigraphy

- Principle of Original Horizontality: Layers are originally deposited horizontally.
- Principle of Superposition: Younger layers lie above older layers.
- Principle of Lateral Continuity: Layers extend laterally in all directions until they thin or terminate.
- Principle of Cross-Cutting Relationships: Features that cut through rocks are younger than the rocks they cut.

## Tools and Data Used in Correlation

- Lithology (rock type and texture)
- Fossil content (biostratigraphy)
- Geochemical signatures
- Structural features
- Radiometric dating results
- Paleomagnetic data

## Methodologies for Rock Correlation in Lab Activities

### Visual and Lithological Matching

- Comparing rock color, grain size, mineral composition, and bedding features.
- Constructing detailed logs of each stratigraphic section.
- Using standardized notation for lithology descriptions.

### Biostratigraphy

- Identifying and comparing fossil assemblages.
- Utilizing index fossils to establish relative ages.
- Recognizing biozones that span multiple sections.

### Geochemical and Mineralogical Correlation

- Analyzing chemical signatures or mineral assemblages.
- Using isotope ratios or elemental analyses to match units.

### Structural and Geophysical Data

- Recognizing structural features like faults or folds that influence stratigraphy.
- Employing seismic or ground-penetrating radar data when available.

### Constructing Correlation Charts and Cross-Sections

- Developing correlation tables that list units across different sections.
- Drawing cross-sections to visualize lateral relationships.

## **Common Challenges in Rock Correlation Lab Activities**

### **Complex Lithological Variations**

- Lateral facies changes can obscure correlations.
- Diagenetic processes may alter original lithology.

### **Fossil Discontinuities**

- Fossil assemblages may be absent, reworked, or poorly preserved.
- Biostratigraphic resolution can be limited.

### **Structural Deformations**

- Folds, faults, and tilting can complicate stratigraphic relationships.
- Recognizing true lateral equivalents versus structural discontinuities.

### **Data Limitations**

- Incomplete exposure or core data.
- Inconsistent sampling density.

### **Subjectivity in Interpretation**

- Different geologists may interpret the same data differently.
- The necessity for standardized criteria.

## **Best Practices for Effective Rock Correlation**

### **Comprehensive Data Collection**

- Collect detailed lithological descriptions.

- Record fossil data meticulously.
- Use geochemical analyses where possible.

## Standardization and Documentation

- Maintain consistent terminology.
- Document all observations and assumptions.

## Use of Multiple Correlation Methods

- Combine lithological, biostratigraphic, and geochemical approaches.
- Cross-validate correlations to increase confidence.

## Critical Evaluation

- Question anomalies or discrepancies.
- Consider structural influences or post-depositional alterations.

## Utilization of Technology

- Employ digital databases and software for correlation.
- Use GIS tools for spatial visualization.

## Sample Questions and Typical Answers in Lab Activities

### Question 1:

Identify the key lithological features that suggest two rock units are equivalent in different sections.

### Sample Answer:

The two units share similar lithology characterized by fine-grained sandstone with cross-bedding, consistent color (light gray), and matching mineral composition (quartz and feldspar). Additionally, both contain similar fossil assemblages of the same index fossil, indicating they are likely lateral equivalents of the same depositional environment.

### Question 2:

How does fossil content assist in correlating stratigraphic units?

Sample Answer:

Fossil content provides biostratigraphic markers that help establish relative ages of rock units. The presence of specific index fossils, which are known to occur within narrow geological time frames, allows geologists to match units across different locations. For example, the discovery of Trilobite sp. in two sections suggests they are from the same or similar time period, aiding correlation.

Question 3:

What are some limitations of using lithology alone for correlation?

Sample Answer:

Lithology alone can be misleading because similar rock types can occur in different depositional environments or may have been altered post-depositionally. Facies changes can cause lateral variations in lithology, potentially leading to incorrect assumptions about equivalence. Therefore, lithological correlation should be supplemented with fossil, geochemical, or structural data.

## Interpreting and Validating Correlation Answers

- Consistency Checks: Ensure that multiple lines of evidence (lithology, fossils, geochemistry) agree.
- Cross-Section Validation: Confirm that the correlation makes sense in the context of the regional stratigraphy.
- Uncertainty Assessment: Acknowledge areas where data are sparse or ambiguous, and suggest further investigation.
- Correlation Confidence Levels: Assign levels (e.g., high, moderate, low confidence) based on the strength of evidence.

## Conclusion and Final Thoughts

The lab activity on rock correlation answers is a vital exercise in practical geology, honing skills in observational analysis, critical thinking, and integrating diverse data sources. Mastery of correlation techniques enables geologists to reconstruct past environments, delineate stratigraphic boundaries, and make informed decisions in exploration and environmental management.

Key Takeaways:

- Correlation is a multi-faceted process requiring meticulous data collection and interpretation.
- Combining lithological, fossil, geochemical, and structural data yields the most reliable results.
- Recognizing limitations and uncertainties is essential for credible interpretations.
- Continuous practice and critical evaluation improve correlation skills over time.

Engaging deeply with lab activities and understanding the rationale behind answers fosters a robust foundation in stratigraphy and enhances the ability to address real-world geological challenges effectively.

Remember: Accurate rock correlation is the cornerstone of geological mapping and resource exploration. Developing proficiency in interpreting lab data and correlating rocks across different sections is indispensable for aspiring geologists and earth scientists.

**Question** What is the purpose of the lab activity on rock correlation? The purpose is to understand how different rock layers can be matched across various locations to determine their relative ages and geological history. How do geologists use rock correlation to interpret Earth's history? Geologists compare rock layers, fossils, and mineral compositions across sites to establish a timeline of geological events and reconstruct past environments. What tools or methods are typically used in rock correlation activities? Tools include stratigraphic charts, fossil identification, mineral analysis, and sometimes geophysical methods like seismic surveys to correlate layers across distances. What are some common challenges faced during rock correlation? Challenges include disrupted layers due to erosion or tectonic activity, similar appearances of different layers, and the difficulty in identifying unique fossils or mineral markers. How do fossils aid in correlating rock layers in lab activities? Fossils serve as index fossils, helping to identify and match layers of the same age across different locations, thereby aiding in accurate correlation. What are the key concepts students should understand from a rock correlation lab activity? Students should understand stratigraphy principles, the importance of fossil evidence, the methods of matching layers, and how correlation helps interpret Earth's geological history.

**Related keywords:** rock activity, correlation answers, geology lab, mineral identification, rock classification, lab exercises, sedimentary layers, geological samples, lab report, earth science activities

## Pogil The Activity Series Answers

**pogil the activity series answers** are an essential resource for students and educators seeking to understand the reactivity of metals and the principles behind chemical reactions. The POGIL (Process Oriented Guided Inquiry Learning) activity series is designed to foster active learning

through guided inquiry, helping learners grasp complex concepts related to the activity series of metals. This article provides a comprehensive overview of the activity series, offers detailed answers to typical POGIL activities, and explains how understanding these answers can enhance your grasp of chemical reactivity and metal displacement reactions.

## Understanding the Activity Series of Metals

### What Is the Activity Series?

The activity series of metals is a ranked list that arranges metals based on their reactivity with other substances, especially acids and water. Metals higher in the series are more reactive and can displace metals lower in the series from their compounds. This concept is fundamental in predicting the outcomes of various chemical reactions involving metals.

### Purpose of the Activity Series

- To predict whether a metal will react with acids or water.
- To determine the feasibility of displacement reactions.
- To understand the relative reactivity of different metals.
- To assist in identifying safe handling and storage of metals.

### Commonly Used Reactivity Series

The activity series typically includes metals such as:

- Potassium (K)
- Sodium (Na)
- Calcium (Ca)
- Magnesium (Mg)
- Aluminum (Al)
- Zinc (Zn)
- Iron (Fe)
- Lead (Pb)
- Copper (Cu)
- Mercury (Hg)
- Silver (Ag)
- Gold (Au)

## Key Concepts in POGIL Activities on the Activity Series

## Displacement Reactions

Displacement reactions occur when a more reactive metal displaces a less reactive metal from its compound. For example:

- Zinc reacting with copper sulfate:  $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$

## Reactions with Water and Acids

Metals react differently depending on their position in the activity series:

- Highly reactive metals (e.g., potassium, sodium) react vigorously with water.
- Moderately reactive metals (e.g., zinc, iron) react slowly or not at all with water but react with acids.
- Less reactive metals (e.g., copper, silver) usually do not react with acids or water.

## Corrosion and Protection

Understanding the activity series helps in preventing corrosion:

- Sacrificial protection involves placing a more reactive metal (like zinc) to protect less reactive metals (like iron).

## Common POGIL Activity Series Questions and Answers

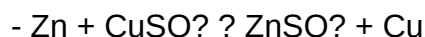
**Question 1: Rank the following metals from most reactive to least reactive: magnesium, copper, zinc, and silver.**

- Most reactive: Magnesium
- Zinc
- Copper
- Least reactive: Silver

Answer Explanation: Magnesium is higher in the activity series, meaning it reacts more readily with acids and water. Zinc is below magnesium but above copper and silver, which are much less reactive.

### Question 2: Will zinc displace copper from copper sulfate? Explain your answer.

Answer: Yes, zinc will displace copper from copper sulfate because zinc is more reactive than copper according to the activity series. The reaction is:

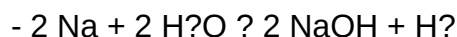


### Question 3: Which metals will react with water? Provide examples from the activity series.

Answer: Metals above hydrogen in the activity series react with water:

- Potassium (K)
- Sodium (Na)
- Calcium (Ca)
- Magnesium (Mg)

Example: Sodium reacts vigorously with water:



### Question 4: Why does gold not react with acids or water? Use the activity series to support your answer.

Answer: Gold is at the bottom of the activity series, indicating it has very low reactivity. Therefore, it does not react with acids or water because it is less reactive than hydrogen and cannot displace other elements from their compounds.

## How to Use the Activity Series in Chemistry

### Predicting Reactions

To predict if a displacement reaction will occur:

- Check the positions of the metals involved.
- If the metal you are adding is higher in the series than the metal in the compound, a reaction will

likely occur.

## Designing Experiments

- Select metals based on their position in the activity series.
- Observe reactions with acids, water, or other metal compounds.
- Record and analyze results to reinforce understanding.

## Safety Considerations

- Highly reactive metals like potassium and sodium should be handled with care.
- Reactions involving these metals should be performed under supervision and with proper safety equipment.

## Summary of Key Points for POGIL Activity Series Answers

- The activity series ranks metals based on reactivity.
- More reactive metals can displace less reactive metals from their compounds.
- Metals react differently with water and acids depending on their position.
- Gold and platinum are least reactive, often unreactive with acids and water.
- Understanding the activity series aids in predicting reactions, preventing corrosion, and designing chemical experiments.

## Additional Tips for Mastering the Activity Series

### Memorization Strategies

- Use mnemonic devices to remember the order of metals.
- Create flashcards with metal names and their positions.
- Practice predicting reactions regularly.

### Real-World Applications

- Corrosion prevention in infrastructure.
- Metal extraction and refining processes.
- Designing batteries and electrochemical cells.

- Understanding environmental impacts of metal waste.

## Practice Questions for Reinforcement

- Which metals can react with acids but not with water?
- How does the activity series explain the corrosion of iron?
- What is the significance of the activity series in recycling metals?

## Conclusion

Understanding the **pogil the activity series answers** is fundamental to mastering concepts in chemistry related to metal reactivity. By exploring the principles behind the activity series, students can predict chemical reactions, understand displacement reactions, and appreciate the importance of metal reactivity in industrial processes and everyday life. Whether you're preparing for exams, conducting experiments, or simply seeking to deepen your understanding of chemistry, mastering the activity series through guided inquiry activities like POGIL can significantly enhance your knowledge and analytical skills.

Keywords: pogil activity series answers, metal reactivity, displacement reactions, activity series of metals, chemical reactivity, corrosion prevention, metal reactions with water, metal reactions with acids, POGIL activities, chemistry education

Pogil the Activity Series Answers: A Comprehensive Guide to Understanding and Using the Activity Series in Chemistry

Understanding the Pogil the Activity Series Answers is essential for students and educators who aim to master the concepts of reactivity in chemistry. The activity series, often presented in Pogil (Process Oriented Guided Inquiry Learning) activities, provides a systematic way to predict the outcomes of single replacement reactions and to understand the relative reactivity of metals and non-metals. This guide aims to demystify the activity series, explain how to interpret Pogil activities related to it, and provide strategies for mastering the answers effectively.

What is the Activity Series?

The activity series is a ranked list of elements (mainly metals) based on their ability to displace other elements from compounds, particularly in aqueous solutions. It helps chemists predict whether a certain single displacement reaction will proceed spontaneously.

Key Features of the Activity Series:

- Reactivity ranking: Elements are ordered from most reactive to least reactive.
- Displacement capability: More reactive metals can displace less reactive metals from their compounds.
- Standard reduction potentials: The series correlates with the elements' tendencies to gain electrons (be reduced).

### The Role of Pogil Activities in Learning the Activity Series

Pogil activities are designed to facilitate inquiry-based learning. When applied to the activity series, these activities encourage students to analyze data, interpret trends, and develop reasoning skills related to chemical reactivity.

#### Typical Pogil Activity Structure:

- Initial exploration: Students observe and analyze experimental data or reaction patterns.
- Guided questions: Foster critical thinking about the relationships between element reactivity and their positions in the series.
- Application exercises: Students predict the outcomes of reactions based on the series and verify their reasoning.

### How to Approach Pogil the Activity Series Activities

Mastering Pogil activities related to the activity series involves a strategic approach:

#### Step 1: Review the Activity Series Chart

Familiarize yourself with the standard activity series chart. Recognize the order of metals and how it relates to their reactivity.

#### Step 2: Understand Key Concepts

- Displacement reactions: Metal A displaces metal B if A is above B in the series.
- Reaction spontaneity: Reactions proceed spontaneously if the element doing the displacing is more reactive.

#### Step 3: Analyze Experimental Data

Look at the data provided in the activity:

- Which reactions occurred?
- Which did not?
- What patterns emerge?

#### Step 4: Answer Guided Questions Carefully

Reflect on the reasoning behind each answer:

- Why does a particular reaction happen or not?
- How does the position of an element in the series influence its reactivity?

#### Step 5: Practice with Variations

Try predicting the outcomes of reactions not explicitly covered in the activity to reinforce understanding.

#### Common Questions and Answer Strategies for Pogil the Activity Series

Below are typical questions encountered in Pogil activities and strategies for answering them effectively.

Question 1: Which metals can displace hydrogen from acids?

Answer approach:

- Refer to the activity series to identify metals above hydrogen.
- Metals like zinc, magnesium, and iron are above hydrogen and can displace it from acids.
- Metals below hydrogen (e.g., copper, silver) cannot.

Question 2: Why does a reaction occur between zinc and copper sulfate but not between copper and zinc sulfate?

Answer approach:

- Look at the position of zinc and copper in the activity series.
- Zinc is above copper, so zinc can displace copper from its sulfate.
- Copper is below zinc, so it cannot displace zinc from its sulfate.

Question 3: Predict whether a given reaction will occur.

Answer approach:

- Check the relative reactivity of the metals involved.
- Confirm whether the metal attempting to displace is higher in the activity series.

Tips for Mastering the Activity Series and Pogil Answers

- Memorize the basic activity series: While not always necessary to memorize the entire series, knowing the general order helps in quick reasoning.
- Understand the underlying principles: Recognize that the activity series is based on oxidation-reduction tendencies.
- Practice with real examples: Use textbook problems, online quizzes, and lab experiments to reinforce your understanding.
- Use visual aids: Create flashcards or charts to visualize the series.
- Discuss with peers or instructors: Clarify doubts and explore different reasoning approaches.

Real-World Applications of the Activity Series

Understanding the activity series extends beyond academic exercises and into practical applications:

- Corrosion prevention: Selecting appropriate metals for construction to prevent rust.
- Electrochemical cells: Designing batteries based on the reactivity of metals.
- Metal extraction: Choosing suitable metals for reduction processes.

Conclusion

Mastering Pogil the Activity Series Answers is a fundamental step toward understanding chemical reactivity and predicting reaction outcomes. By familiarizing yourself with the structure of the activity series, practicing interpretation of Pogil activities, and applying critical reasoning strategies, you will enhance your comprehension of oxidation-reduction processes. Remember, the activity series is more than just a list—it's a powerful tool that unlocks the predictive power of chemistry in both academic and real-world contexts.

Empower your chemistry learning journey by engaging deeply with Pogil activities and the activity series. With practice and understanding, you'll develop the confidence to analyze reactions critically

and apply your knowledge effectively.

**Question** What is the purpose of the POGIL activity series activity? The purpose is to help students understand the reactivity of metals by arranging them in a activity series based on their ability to displace or react with other substances. How do I determine the position of a metal in the activity series using POGIL activities? You determine the position by analyzing experimental results from displacement reactions and comparing how easily metals react with acids, water, or other solutions, as guided by the POGIL activity instructions. Are the answers to the POGIL activity series activity available online? Yes, various educational resources and teacher websites provide answer keys to POGIL activity series activities, but it's best to use them to understand concepts rather than just copying answers. What are common mistakes students make when completing the POGIL activity series activity? Common mistakes include misinterpreting experimental data, confusing the reactivity order of similar metals, or not following the experimental procedure carefully as outlined in the activity guide. How can I use the POGIL activity series to predict chemical reactions? By understanding the activity series, you can predict whether a metal will displace another from a compound or react with acids, based on their positions in the series. Is the POGIL activity series activity suitable for beginner chemistry students? Yes, it is designed to introduce students to the concept of reactivity and series of metals in a collaborative and inquiry-based manner suitable for beginners. Where can I find detailed answers or explanations for the POGIL activity series? You can find detailed explanations in teacher guides, educational websites, or textbooks that accompany the POGIL curriculum, and some online forums or study groups discuss these activities. How does understanding the activity series help in real-world applications? Understanding the activity series helps predict corrosion, metal extraction, and reactivity in industrial processes, making it essential for chemistry-related industries. Can I complete the POGIL activity series activity without prior knowledge of reactivity series? While prior knowledge helps, the activity is designed to guide students through the process of discovering the reactivity series through experimentation and guided inquiry. What should I do if I get stuck on the POGIL activity series question? Review the experimental data, revisit the activity instructions, collaborate with classmates, or consult your teacher for hints to help clarify the concepts and move forward.

**Related keywords:** activity series, pogil worksheet, reactivity series, metal reactivity, pogil activities, activity series chart, metal reactivity order, chemistry pogil, activity series answers, metal displacement reactions

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